Balancing Chemical Equations Student Guide Name: \_\_\_\_\_\_\_\_\_\_\_

**Start:**  1.

Google: PhET Balancing Chemical Equations

 2. Click on the first link

1. Click on the  button.
2.  Explore the simulation. Be sure to click on everything.
3. Try both 
	1. Talk about how they may be useful in the activity.

 ![MC900432579[1]]()Tell me what the scale and bar charts are helping you do? What is the purpose of these tools?

\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

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Which one does your group prefer **and** why? \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

* 1. Fill in the box-



6. ![MC900432579[1]]()Is this equation balanced? Explain. \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

Change the coefficients in the equation until you get a !

What does this smiley face mean?

\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

 **Teacher check-off**

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7. ![MC900432579[1]]()Is this equation balanced? Explain. \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

Using the simulation- add coefficients so that the equation is balanced correctly.

\_\_\_\_\_\_ H2O \_\_\_\_\_\_ H2 + \_\_\_\_ O2

8. 

Using the simulation- add coefficients so the equation is balanced correctly.

\_\_\_ CH4 + \_\_\_O2 \_\_\_ CO2 + \_\_\_\_ H2O

 List all reactants: List all products:

\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

9. Discuss the maximum (highest number) of ammonia molecules the simulation will let you make? Draw them:
![MC900432579[1]]()In the product, ammonia (2NH3), What does the (2) tell you about the equation?

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What does the (3) tell you?\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

 **Teacher check-off/check-in… Go on to  tab...**